

A New 15,16-Dimethyl-30-glyceryloxytriacontanoic Acid from Lipids of *Thermotoga maritima*

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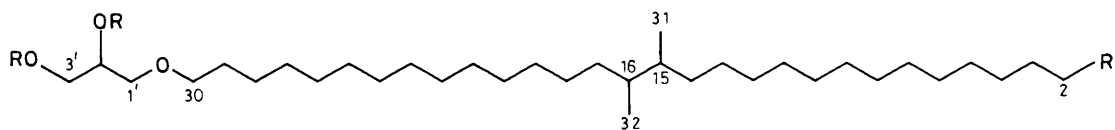
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The lipids of the thermophilic anaerobic eubacterium *Thermotoga maritima* are mainly based on: (a) n-C₁₄, n-C₁₆, and n-C₁₈ fatty acids; (b) 15,16-dimethyltriacontanedioic acid (diabolic acid); (c) a new compound identified by chemical and degradative evidence as 15,16-dimethyl-30-glyceryloxytriacontanoic acid.

Here we report data on lipid structures of *Thermotoga maritima*, a new anaerobic eubacterium growing at temperatures up to 90 °C, isolated from geothermally heated sea floors in Italy and the Azores.¹ Lyophilized cells afforded a 7–8% yield of lipids, which were refluxed with methanol/HCl (9:1) for 6 h. The CHCl₃-soluble fraction of the methanolysis mixture was chromatographed on a silica-gel column. Light petroleum/Et₂O (99:1) eluted a mixture of fatty acid methyl esters (ca. 36% of the CHCl₃-soluble fraction) identified by g.l.c. as methyl myristate (12%), palmitate (86%), and stearate (1%); light petroleum/Et₂O (98:2) eluted a long-chain dicarboxylic dimethyl ester (ca. 33% of the CHCl₃-soluble fraction); CHCl₃ eluted a novel compound (**1a**) (ca. 30%) and trace amounts of an identified compound; CHCl₃/MeOH (80:20) gave a more polar, and as yet unidentified compound (1%).

The long chain dicarboxylic dimethyl ester [*R_f* ca. 0.4 on t.l.c., silica gel in light petroleum/Et₂O (98:2)] was identified as dimethyl 15,16-dimethyltriacontanedioate (trivial name, diabolic acid) by a comparison of its physical properties {[α]_D, ¹H and ¹³C n.m.r. electron-impact mass spectrometry (E.I.M.S.)} with those reported in the literature.²

The new compound (**1a**) [*R_f* ca. 0.5 on t.l.c., silica gel in CHCl₃/MeOH (95:5)] was acetylated with Ac₂O/pyridine (9:1) at 60 °C overnight. The acetylated derivative (**1b**) [*R_f* 0.95 on t.l.c. silica gel, CHCl₃/MeOH (95:5)] was purified by silica gel column chromatography with CHCl₃ as eluant. The E.I.M.S. of diacetate (**1b**), besides *M*⁺ at *m/z* 668 (0.1%), includes diagnostically important peaks at *m/z* 637 (*M*⁺ – CH₃O; 3.1%), 609 (*M*⁺ – OCOCH₃; 4.1), 608 (*M*⁺ – AcOH; 4.1), 595 (*M*⁺ – CH₂OAc; 0.9), 565 (*M*⁺ – AcOH and Ac; 7.2), 548 (*M*⁺ – 2AcOH; 4.3), 535 (*M*⁺ – CH₂OAc and



(**1a**); R = H, R' = CO₂Me

b; R = Ac, R' = CO₂Me

c; R = H, R' = CH₂OH

AcOH; 6.0), 523 (M^+ - CH(OAc)-CH₂OAc; 7.1), 493 [M^+ - C₇H₁₁O₅; C(30)-O cleavage; 3.9], 491 (M^+ - CH₃OH and CHOAc-CH₂OAc; 4.0), 479 [M^+ - C₈H₁₃O₅; C(29)-C(30) cleavage; 1.7], 477 [M^+ - CH₃OH and C₇H₁₁O₄; C(1')-O cleavage; 1.7], 461 [M^+ - CH₃OH and C₇H₁₁O₅; C(30)-O cleavage; 8.7], 339 [M^+ - AcOH and C₁₇H₃₃O₂; C(15)-C(16) cleavage; 32.9], 159 [loss of C₃₃H₆₅O₃; C(1')-O cleavage; 100], thus localizing the ether linkage at the 1'-position of glycerol and *vic*-dimethyl branching at the 15-16 position of the C₃₀ aliphatic chain.

The ¹H n.m.r. spectrum (CDCl₃) of (**1a**) shows in addition to the CH₃O signal (δ 3.67) significant resonances at δ 3.88 (1 H, br.m, H-2'), 3.72 (2 H, br.d, H-3'), 3.53 (2H, d, H-1'), 3.46 (2 H, t, H-30), 2.30 (2 H, t, H-2), 0.73 (6 H, d, H-31 and H-32). The ¹H n.m.r. spectrum of (**1b**) shows singlets due to two acetyl groups at δ 2.06 and 2.08, a doublet at δ 3.54 due to the 1'-protons, and an ABX system centred at δ 4.25 and 5.18 due to the 3' and 2' protons, respectively.

The ¹³C n.m.r. data of (**1a**) fully confirm the 15,16-dimethyl-30-glyceryloxytriacontanoic acid structure. In particular the steric compression shift of methyl resonance to higher field (¹³C n.m.r.: δ 14.5), also observed for diabolic

acid,² indicates a higher probability of *gauche* environment for these methyl groups. Further details on *vic*-dimethyl branching were obtained by chemical degradation of (**1a**). In this procedure (**1a**) was reduced to (**1c**) with NaAlH₂(OCH₂-CH₂OCH₃)₂³ and then converted by HI/LiAlH₄ degradation⁴ in the parent C₃₂ hydrocarbon. This was identified as 15,16-dimethyltriacontane by comparison with E.I.M.S. data previously reported.²

Full characterization of complex lipids is in progress.

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References

- 1 R. Huber, T. A. Langworthy, H. Konig, M. Thomm, C. R. Woese, U. B. Sleytr, and K. O. Stetter, *Arch. Microbiol.*, 1986, **144**, 324.
- 2 R. A. Klein, G. P. Hazlewood, P. Kemp, and R. M. C. Dawson, *Biochem. J.*, 1979, **183**, 691.
- 3 F. Snyder, M. L. Blank, and R. L. Wykle, *J. Biol. Chem.*, 1971, **246**, 3639.
- 4 M. De Rosa, A. Gambacorta, B. Nicolaus, B. Chappe, and P. Albrecht, *Biochim. Biophys. Acta*, 1983, **753**, 249.